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Homogeneous and
Heterogeneous
Mixtures
Examples,
Classification*

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*of Matter,
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Lifewave Webinar
1.1

Pure Substances
and Mixtures,
Elements \u0026
Compounds,
Classification
of Matter,
Chemistry

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Examples, And

3.2.1/3.2.2

Describe the
differences

between Chapter

elements,

compounds and

mixtures. ~~States~~

~~of Matter :~~

~~Solid Liquid Gas~~

~~10 Amazing~~

~~Experiments with~~

~~Water Changes in~~

the State of

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Matter – Science

Experiments –

Solid to Liquid,

Liquid to Gas,

Gas to Liquid

Changes of

States of Matter

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Mass in Chemical

Reactions The

States of Matter

– Changes of

State – Solid

Liquid Gas

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States of Matter
Understanding

*Atoms, elements,
and molecules*

Part #1 (9min)

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A change in

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physical form of
a substance or
physical
properties of a
substance like
color, melting
point, state,
size, shape, and
arrangement of
molecules is
known as
physical change.
A change in
chemical

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properties of a substance like composition, reactivity with water, corrosiveness, and pH is known as chemical change. In chemical change, a new substance will be formed.

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35. Aqueous
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potassium iodide
and silver
nitrate are
mixed, forming
the precipitate
silver iodide.

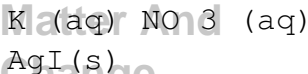
chemical

equation: $KI(aq) + AgNO_3(aq) \rightarrow KNO_3(aq) + AgI(s)$

complete ionic

equation: $K^+(aq) + I^-(aq) + Ag^+(aq) + NO_3^-(aq) \rightarrow K^+(aq) + NO_3^-(aq) + AgI(s)$

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Chemical Reactions Chapter 12

Solution.

Physical:

boiling and

melting are

physical

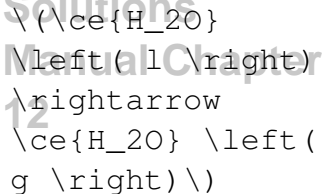
changes. When

water boils, no

bonds are broken

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or formed. The
change could be
written:



Chemical: The
dark grey nail
changes color to
form an orange
flaky substance
(the rust); this

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must be a
chemical change.

3.6: Changes in Matter: Physical and Chemical Changes ...

240 Chemistry:
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Compare and contrast sublimation and evaporation. In both processes, the substances become a vapor. During sublimation, the substance goes from the solid phase directly to the vapor phase. During

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evaporation,
particles in a
liquid gain
enough

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